

## Exercise 2.2-2

### Quick Questions to

#### 2.2 Extrinsic Point Defects and Point Defect Agglomerates

##### 2.2.1 Impurity Atoms and Point Defects; 2.2.2 Local and Global Equilibrium

Here are some quick questions:

- The *answers* are sometimes (and possibly only indirectly) contained in the links.

Let's look at combined defects - double vacancies, impurity atom - vacancy complex (and so on):

- Derive from the mass action law and write down the *essential* equations for the concentrations of

- [Divacancies](#) ( $c_{2V}$ )
- [Clusters](#) with  $n$  vacancies ( $c_{nV}$ ).
- [Impurity atom - vacancy complex](#) ( $c_C$ ;  $c_F$  = impurity atom conc.).

Forget pre-exponential factors etc., if you don't remember, or make simple assumptions.

Now let's do something important. *You really should do this, it will teach you a lot!*

- Make *sketches* of various concentrations in an Arrhenius plot. Try to produce intelligent and neat sketches with parameters as follows:

- The concentration of single vacancies at  $T_m$  is roughly  $c_V \approx 10^{-4}$
- Positive binding energies in all case; about **10 % - 20 %** of the vacancy formation enthalpy  $H_F(V)$ .
- Always include the Arrhenius plot for the single vacancy as reference.

- First*, produce one Arrhenius diagram showing single *and* double vacancies.

- Second*, produce an Arrhenius diagram for single vacancies, impurities, and impurity atom - vacancy complex

- Third*, produce one Arrhenius diagram showing single *and* double vacancies *but* assume that the single vacancy concentration *cannot* decline anymore at some lower temperature.

- Discuss your curves (in particular the **2nd** and **3rd**), take into account how the temperature changes in "theory" and in "real life"

Now a few really quick ones:

- If all vacancies present at thermal equilibrium near the melting point at a concentration of  $c_V \approx 10^{-4}$  end up in vacancy clusters with an average of **100** vacancies, what is the concentration of these clusters? What is their average cluster distance compared to the average vacancy distance (assume a typical lattice constant around **0.3 nm**)?
- Given an equilibrium vacancy concentration of  $c_V$ , an (substitutional) impurity concentration  $c_F$ , and some binding enthalpy and entropy  $H_C$  and  $S_C$ , the concentration  $c_C$  of vacancy - (substitutional) impurity complexes should be proportional to:.....?
- What would you expect for the case of no binding enthalpy and entropy?