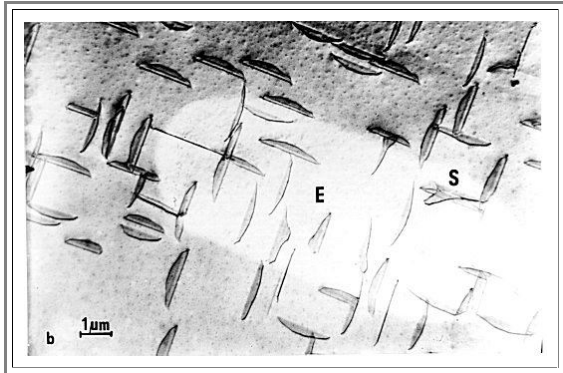


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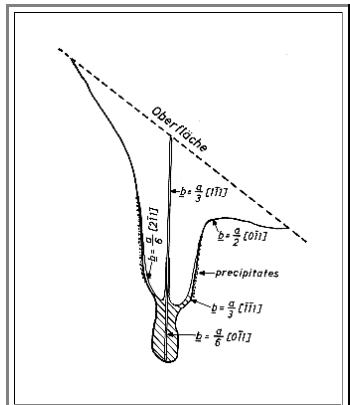
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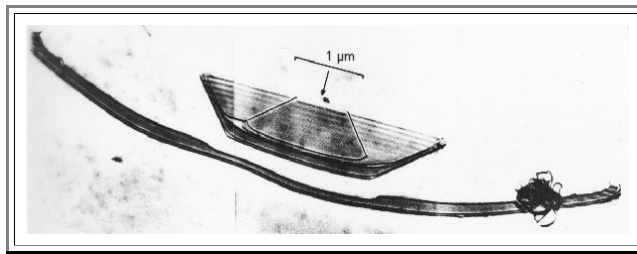


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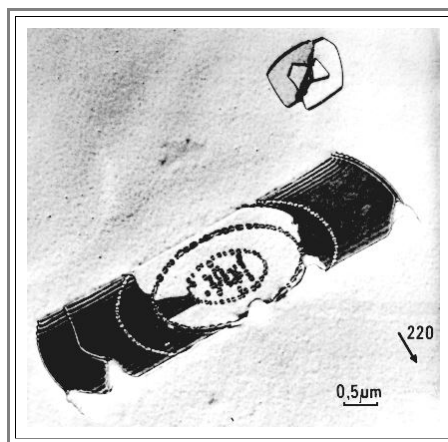
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- ▶ If the **Si** contains some supersaturated oxygen (at high temperatures an equilibrium defect as an interstitial; "**O_i**"), we may observe **internal oxidation**.

 - A **SiO₂** precipitate forms by the agglomeration of **O_i**; but this may equally well be considered to be an internal oxidation of a small volume of **Si**. Again, interstitials are produced with the tendency for agglomeration.
 - In contrast to surface oxidation, nucleation is rather easy. The small **SiO₂** precipitate, especially if it is not spherical, has a stress field that helps to nucleate the stacking fault of the interstitials. We thus find oxide precipitates surrounded by large stacking faults.
- ▶ Both processes - the oxide precipitation and the stacking fault formation - occur simultaneously; new precipitates may be nucleated at the Frank dislocation and vice versa.

 - In the course of several high temperature treatments; the processes start all over again and complicated structures develop:



- Several perfect stacking fault loops overlap (truncated by the sample surface, one of which has been preferentially etched; the etch pits at the dislocations are clearly visible). Some of the loops serves as nucleation sites for a second and third round of oxygen precipitation (shown as small coffee-bean like contrasts).