

Alloying Elements and Properties of Steel

Here are a few major alloying elements for steel and some information about what they can do.

This list is based on the "[Materials in Action Series; Structural Materials](#)"

Illustration

| Element | Influence on Ferrite | Influence on Hardenability | Tendency to form hard Carbides | Major Functions |
|-------------------------|---|---------------------------------------|--------------------------------|---|
| Manganese Mn | Powerful solution strengthener | Moderate increase | Middle | <ol style="list-style-type: none"> 1. Takes care of <i>Sulphur (S)</i>. 2. <i>Cheap</i> increase of hardenability. |
| Silicon Si | Hardens, but <i>reduces ductility</i> | Moderate increase | - | <ol style="list-style-type: none"> 1. Deoxidation of <i>liquid steel</i>. 2. Improves oxidation resistance. 3. Strengthens low alloy steel. 4. Increases <i>electrical resistivity</i> (<i>important for transformer cores</i>). |
| Chromium Cr | Strengthens a little Provides corrosion resistance | Moderate increase | Strong | <ol style="list-style-type: none"> 1. Corrosion resistance. 2. Hardenability. 3. <i>Abrasion resistance</i> (needs high C, too). 4. Strength + oxidation resistance at <i>high T</i>. |
| Titanium Ti | Age hardening possible | Very strong increase | Extremely strong | <ol style="list-style-type: none"> 1. Forms hard carbides. 2. Prevents <i>local depletion</i> of C carbon in stainless steels due to Cr-carbide formation |
| Vanadium V | Moderate solid solution hardening | Very strong increase | Very strong | <ol style="list-style-type: none"> 1. <i>Restricts grain coarsening</i> of austenite. 2. Increases hardenability. 3. Delays softening during tempering. |
| Nickel Ni | Strengthens | Mild improvement stabilizes austenite | - | <ol style="list-style-type: none"> 1. Improves strength and toughness at <i>subzero T</i>. 2. Together with Cr provides austenitic steel. |
| Molybdenum Mo | Age hardening possible | Strong increase | Very strong | <ol style="list-style-type: none"> 1. Increase hardenability. 2. Prevent <i>embrittlement</i> of certain Ni/Cr steels. 3. Keeps strength at higher T. 4. Restricts austenite <i>grain growth</i>. 5. Improves <i>corrosion resistance of stainless steels</i>. 6. Provides carbides with high <i>abrasion resistance</i>. |
| Cobalt Co | Strengthens in solid solution | <i>Decreases</i> slightly | Like Fe | <ol style="list-style-type: none"> 1. Contributes hardness at <i>moderately high T</i>. |

The list could go on for a while, of course. It includes some properties not much discussed before, for example:

- Behavior at low and/or high temperatures.
- Properties like wear (or abrasion) resistance or corrosion resistance (note that stainless steel, while oxidation resistant, might nevertheless corrode in some other chemical environment).
- Making steel in the first place (look for "liquid").
- Counteracting the effects of other elements.
- Keeping the structure from unwanted changes ("grain growth")