

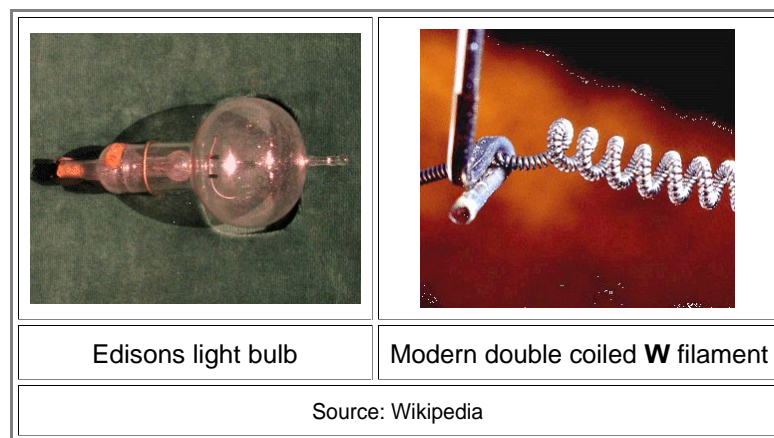
5.3 Optical Components

5.3.1 Light Sources

Conventional Light Sources

There is not much to say about conventional light sources like simple light bulbs, "halogen" light bulbs, gas-discharge sources and so on. You all are quite familiar with them. What follows gives the bare essentials.

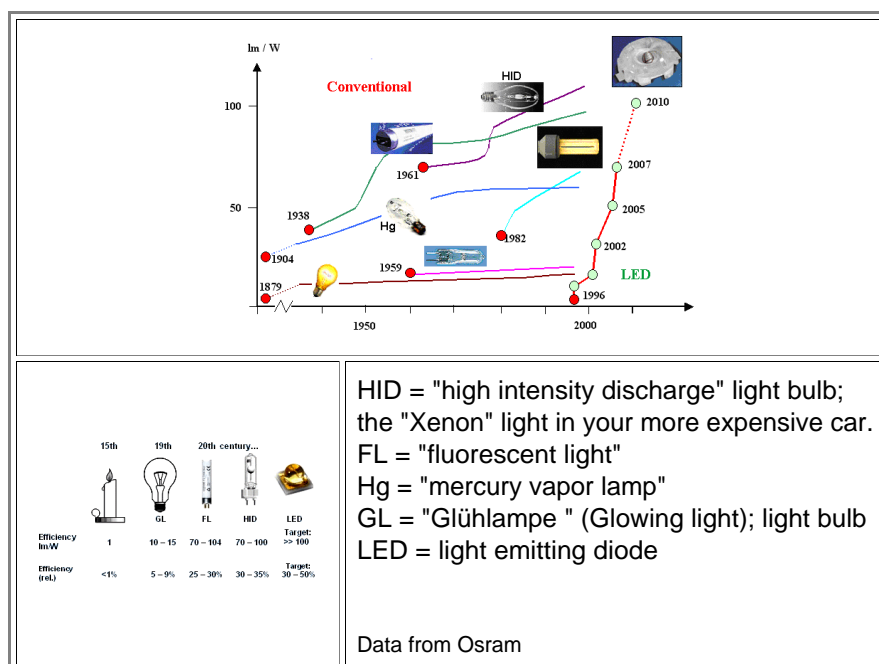
- **Thomas Edison** is usually credited as inventor of the light bulb in **1880** but there were many others working on "light bulbs" as early as **1840**. Edison's breakthrough probably was due to a combination of three factors: an effective incandescent material, a higher vacuum compared to others, and a high resistance that made power distribution at high voltages from a centralized source economically viable.
- Consider a **100 W** bulb operated at **230 V**. It draws **0,44 A** and thus has a resistance of **227 Ω** . This is not easily achieved with the metal wires then available. Edison of course, used carbon. It took until about **1905** before tungsten (**W**) filaments were used and until about **1913** before an inert gas like **N₂** was inside the bulb instead of vacuum.
- In fact, present day light bulbs are high-tech objects despite their lowly image. If you have doubts about this consider: How would you make a "coiled coil" filament as shown below for a standard **1 €** light bulb a from an extremely hard to shape material like **W** in such a way that it is extremely cheap?



It is hard for us to imagine the impact of "easy" light on humankind. Nevertheless, the **120+** years of illumination by **incandescent light** has to come to an end **right now** for reasons **already given**

Fluorescent and gas discharge light sources have better efficiencies (and **efficacies**) than "black body radiators" but are not without problems of their own.

- The pictures tell it all, just look at the LED branch. No more needs to be said about "conventional light sources".

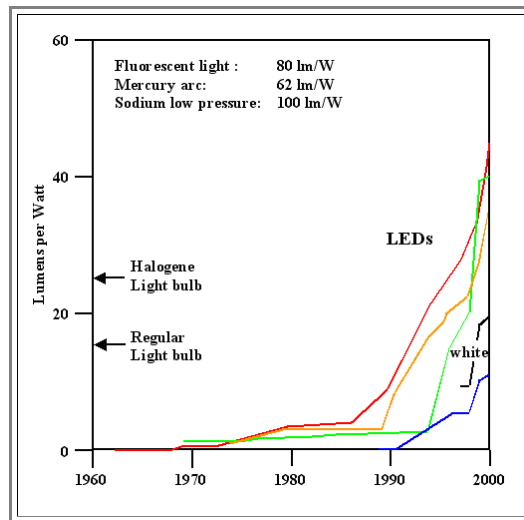


Light Emitting Diode

Light emitting diodes or **LED's** nowadays come in two variants: "Standard" **LED's** made from *inorganic* crystalline semiconductors based on, e.g., **GAAIAs**, **GaP** or **GaN** and "**organic**" **LED's** or **OLED's**.

- **OLED** devices are coming into their own right now (2011). They are not yet mass products for general lightning applications but we will find out how far they will go in the near future (based on the work of possibly you and other materials scientists and engineers; who else?).
- Standard **LED's** have been around for more than 40 years by now. However, they used to be only red in the beginning, see the picture below, and their efficiencies were lousy. The breakthrough came around **1990** when **Shuji Nakamura** of Nichia Corporation almost single-handedly introduced the **GaN** based blue **LED**. This started the ongoing revolution of world wide lighting that will contribute in a major way to saving the planet from the climate crisis. Of course, if you google "Nakamura" you will find a soccer player first.

The picture below gives an idea of what was happening. Nobody seem to have updated this picture but the trends continued. The **LED** market is growing rapidly



- In analogy to "[Moore's law](#)", "**Haitz's Law**" has been proposed: In every decade, the cost per lumen (unit of useful light emitted) falls by a factor of **10**, the amount of light generated per LED package increases by a factor of **20**, for a given wavelength (color) of light. Haitz also predicted that the efficiency of **LED-based** lighting could reach **200 lm/W** (lumen per Watt) in **2020** crossing **100 lm/W** in **2010**.

- **This is important:**

More than **50%** of the electricity consumption for lighting or **20%** of the totally consumed electrical energy would be saved reaching **200 lm/W**

- So get going, young Material Scientist!

What does one need to do to make better (and cheaper) **LED's**?

- As a first step you must learn a minimum about [semiconductor physics](#) or [Halbleiterphysik](#) and [semiconductor technology](#).
- The links provide starting points because *we are not going to do that here*.

Laser

All the light sources discussed so far share certain broad characteristics:

- They emit either a whole **spectrum**, i.e. light with many colors, several spectral lines, or in the case of **LED's** only one line but with a rather large half-width.
- Their light may come from a small area ("point source"; e.g. standard LED), from a longish area ("fluorescent tubes") or even from a large area (OLED's) and cannot really be processed into that **parallel beam** always used for illustrating optical stuff
- The light is emitted in **many directions** with various characteristics but **never** in only one direction.
- The light is never fully coherent and mostly **rather incoherent**.
- The light is mostly not **polarized**

Negate everything in that list (except, maybe, polarization) and you have a **Laser**, a device that operates on the principle of **L**ight **A**mplification by **S**timulated **E**mission of **R**adiation.

Lasers are rather recent light sources; the first one was built by **Maiman** in **1960**; for a short history use the [link](#)

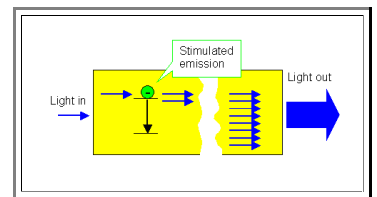
We cannot go much into the principles of Lasers here. We only look at a few basic concepts and keywords..

The name "LASER" says it all. To understand the very basic principles of Lasers, we look at a sequence of a few simple pictures

First we need **L**ight **A**mplification. For that we need a material with two suitable energy levels, $\Delta E = h\nu$ apart. Light results whenever the electron jumps from the higher level to the lower (ground) level one with a basic frequency of ν Hz. Note that this is not true for just any levels; the electron may get rid of its energy in other ways, too, e.g. in indirect semiconductors.

Second we need **stimulated emission**, a phenomenon that was calculated and predicted by **Albert Einstein** in **1916**. In simple terms, stimulated emission means that a photon with the energy ΔE , when encountering an electron sitting on the upper energy level, stimulates it to "fall down" and to emit a photon that is identical in wave vector, and phase to the one that stimulates the process (and does not get absorbed!)

Instead of one photon we have now two identical one. We have achieved **light amplification**. The two photons now stimulate other electrons along their way to produce more photons, all being **fully coherent**.. A lot of light now merges from the output.



The process from above, however, only works once - until all electrons that happens to populate the upper energy level are "down".

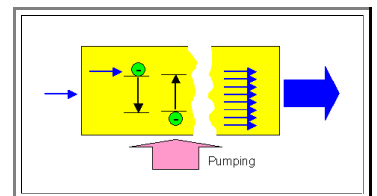
For a material with a dimension of $l = 1 \text{ cm}$ this takes about $t = \tau_{\text{mat}} // \approx [1/(2 \cdot 10^9)] \text{ s} = 0,5 \text{ ns}$, so we would have a rather short light flash.

or a "**cw**" or **continuous wave Laser** we obviously need to kick the electrons up to the higher energy level - just as fast as they come down - by "**pumping**" the Laser. In fact, we need to have more electrons sitting at the high energy level all the times than at the lower level. This is a very unusual state for electrons called **inversion**.

Pumping requires that we put plenty of energy into the system all the time. This can be done by intense illumination (obviously with light of somewhat higher energy than ΔE). Some Lasers of the US military were supposed to be pumped by **X-rays** produced by a nuclear explosion (no joke). They would not live long but still be able to produce a short-lived ultra-high intensity Laser beam suitable for shooting down missiles.

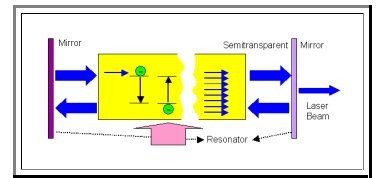
Our cheap, simple and long-lasting semiconductor lasers, in contrast, are "simply" pumped by running a **very large** current density ($> 1000 \text{ A/cm}^2$) through a suitable **pn-junction** in some **direct** semiconductors. [This link](#) gives an idea of what that means.

Note that the incoming photon could just as well kick a lower electron up, than it would be **absorbed**. The photon generated at random some time later when the electron moves back down again is **not** adding to the desired output, it just adds noise.



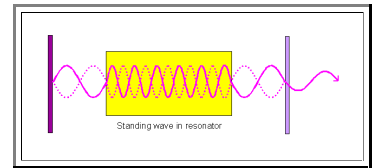
We are not done yet. The picture above are greatly simplified because in reality we would produce light beams running in all kinds of directions. That's not what we want.

- Just as important, the energy of the light produced would not be exactly ΔE but, roughly, $\Delta E \pm kT$ since our excited electrons would also have some thermal energy. For a good monochromatic light, an energy or frequency spread of about **1/40 eV** at room temperature is ridiculously large, so we must do something.
- What we do is putting the pumped material inside a "Fabry Perot" **resonator**. This is nothing else but two mirrors (one with a reflectivity less than **100 %**, i.e. "semi" transparent) that are exactly parallel (within fractions of a μm) and at a distance **L** from each other.
- The light generated then is reflected back and forth. For reasons [clear to us now](#), only waves with $\lambda = 2L/m$; **m = 1,2,3,... will "fit"**.
- A certain part of the light impinging on the "semi" transparent mirror leaks out, forming our .now fully monochromatic and coherent Laser beam. It propagates in one direction only (here perpendicular to the mirrors).

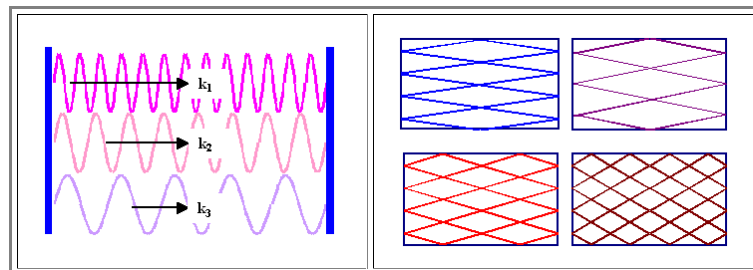


The way to visualize that is shown here.

- We have one standing wave right between the two mirrors. Note that the wave length in the material is different from that in air; you must take that into account when going through numbers.
- Note that the picture for an organ pipe with an acoustic wave inside would, in principle look exactly the same. The pipe would leak some of the wave and you hear a tone with a well defined frequency.
- This looks pretty involved, so how come that we have ultra-cheap Lasers in DVD drives? Because you don't need extra mirrors, you just use the internal surface of your semiconductor single crystal that reflect parts of the beam according to the [Fresnel equations](#). If you obtain those surfaces by cleaving down a low-index plane, they are automatically exactly plane parallel. That makes Lasers more **simple**.
- However, typically Lasers are far more complicated than shown here



An organ pipe or any longish musical instrument will not only produce a tone with one frequency ν_0 but also the harmonics or overtones **m · ν_0** . Same for our Laser, of course, as shown below left.



- A musical instrument that isn't long and slender like an organ pipe or a flute (i.e. an essentially 1-dim. system) but a rectangular box (or a complex-shaped body shape like a violin, can contain standing waves in all directions with many possible wavelengths. Same for our Laser; cf.. the situation in the figure on the upper right.
- So depending on the exact shape of the laser, the way it's pumped, and so on and so forth, there can be more than just one **Laser mode**
 - We needed to get to that word. so let's repeat: There can be more than just one standing wave inside a Laser resonator, or a real laser might emit more than just **one** mode.
- We will not discuss what kinds of Lasers we find for all kinds of applications here. There is a bewildering variety and more and more different kinds are introduced. We just note one important item:
 - Increasing the frequency / energy of Lasers becomes "exponentially" difficult because with increasing photon energy the number of ways it can be absorbed increases rapidly (there are lot of empty states far above some densely populated ground level onto which electrons could be "kicked") but only one state is useful for lasing!
 - That's why there aren't so many **UV** Laser around and no **X-ray** Lasers yet.